

¹³C NMR ANALYSIS OF ALKALOIDS FROM *PESCHIERA FUCHSIAEFOLIA*

RAQUEL M. BRAGA, HERMOGENES F. LEITÃO FILHO* and FRANCISCO DE A. M. REIS†

Instituto de Química, Universidade Estadual de Campinas, C.P. 6154, Campinas 13100, São Paulo, Brazil

(Received 8 March 1983)

Key Word Index—*Peschiera fuchsiaefolia*, Apocynaceae, bisindole alkaloids, voacamine, decarbomethoxyvoacamine, demethylvoacamine, dihydrovoacamine, voacamidine, affinisine, 16-epiaffinisine, pervine, voachalotine, voacangine, voacanginehydroxyindolenine, ¹³C NMR

Abstract—Fractionation of an ethereal extract of *Peschiera fuchsiaefolia* resulted in the isolation of decarbomethoxyvoacamine, demethylvoacamine, voacamidine, pervine, 16-epiaffinisine and voacanginehydroxyindolenine, together with the previously reported alkaloids voacamine, voacangine, voachalotine and affinisine. Analysis of the ¹³C NMR spectra of the bisindole alkaloids and of 16-epiaffinisine is reported.

INTRODUCTION

Reports on the anticancer activity of bisindole alkaloids of the voacamine type [1] led us to re-investigate the bark of *Peschiera fuchsiaefolia* [2, 3], directing our research on the isolation of these compounds.

RESULTS AND DISCUSSION

After alkalization the ground bark of *P. fuchsiaefolia* was extracted with ether. The crude extract was dispersed in 10% acetic acid and the aqueous solution was extracted with chloroform and ether at different pHs, yielding fractions A, B and C. These fractions were monitored by mass spectrometry, which revealed that B was rich in dimeric compounds. This fraction was then submitted to Sephadex LH 20 column chromatography allowing a good MW separation, followed by preparative TLC purification leading to the isolation of the bisindoles voacamine (1a), decarbomethoxyvoacamine (1b), demethylvoacamine (1c) and voacamidine (1d), and the monomeric indoles affinisine (2) and voachalotine (3). Pure voacangine (4), voacanginehydroxyindolenine (5) and voacamine (1a) were obtained from fraction A, while fraction C furnished two α -acylindoles, pervine (6) and 16-epiaffinisine (7), which were not reported previously [2, 3].

The ¹³C NMR spectra were interpreted on the basis of standard chemical shift theory, comparison with reference compounds and mainly by analysis of the SFORD and fully coupled spectral data.

Table 1 presents the ¹³C NMR data of compounds 1a–1f. The shift assignments were made in conformity with the published data of voacangine (4) [4, 5], vobasine (8) [6] and ibogaine (9) [4, 5]. The replacement of the C-3 hydroxyl group of 8 by a voacangine unit in voacamine (1a) produced the expected shielding at C-3 ($\Delta\delta$ 29.4) and

deshielding at C-14 and C-15 ($\Delta\delta$ 1.1 and 4.1, respectively). Analogous effects were observed in the tabernaeanalines [7]. On the other hand, in the voacangine moiety the replacement of the hydrogen at C-11 by a vobasine unit induced a deshielding of the *ipso* ($\Delta\delta$ 15.7) and shielding of the *ortho* carbons ($\Delta\delta_{C-10}$ 3.0 and $\Delta\delta_{C-12}$ 1.4). The remaining carbons showed little or no modification.

The R configuration at C-20 of dihydrovoacamine (1e), the sole product of the catalytic hydrogenation of 1a, was deduced from the chemical shifts of C-14 and C-16. C-14 at δ 31.4 was shielded ($\Delta\delta$ 5.2) in 1e by comparison with the same site of 1a due to a γ -interaction with C-19, while C-16 was deshielded ($\Delta\delta$ 3.7). (The differences produced by the α or β ethyl group orientation were previously discussed for tabernaeanalines A and B [7], dregamine and tabernamontanine [6].)

Comparison of the ¹³C chemical shifts of demethylvoacamine (1c) and voacamine (1a) revealed some interesting conformational aspects concerning the *N*₆-methyl group. Introduction of the *N*-methyl group into 1c induced simultaneous shielding at C-6 ($\Delta\delta$ 4.8) and C-16 ($\Delta\delta$ 6.3). These facts can be explained by taking the inversion of *N*₆ into consideration. Thus the rapid interconversion between the equatorial and axial *N*₆-methyl group leads to the observed shieldings. Analogous effects were observed at C-6 ($\Delta\delta$ 6.3) and C-16 ($\Delta\delta$ 2.6) of 1e for the introduction of the *N*-methyl group in demethyldihydrovoacamine (1f) (going from 1f to 1e). The larger $\Delta\delta$ value at C-6 suggests a preferential equatorial position for the *N*₆-methyl group in the dihydro compound, thus avoiding an additional methyl H₂₀ 1,3-diaxial interaction.

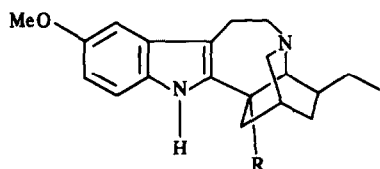
An interesting feature revealed in our work was the isolation of demethylvoacamine (1c) together with voacangine (4) and the α -acylindole pervine (6), implying that both monomers 4 and 6 are precursors in the biosynthetic pathway of 1c. Though *in vitro* data [8] support the above suggestion, rigorous *in vivo* experiments would provide final confirmation.

EXPERIMENTAL

Mps are uncorr. Specific rotations were measured in CHCl₃, UV spectra in EtOH and IR spectra in CHCl₃. ¹H NMR spectra

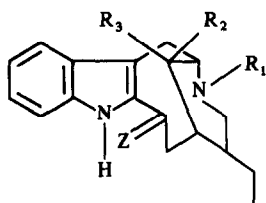
* Present address: Departamento de Morfologia e Sistemática Vegetal, Instituto de Biologia, Universidade Estadual de Campinas, Campinas 13100, São Paulo, Brazil.

† To whom correspondence should be addressed.



4 R = COOMe

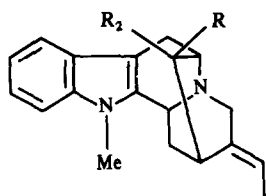
9 R = H



6 R₁ = H, Z = O, R₂ = H, R₃ = COOMe

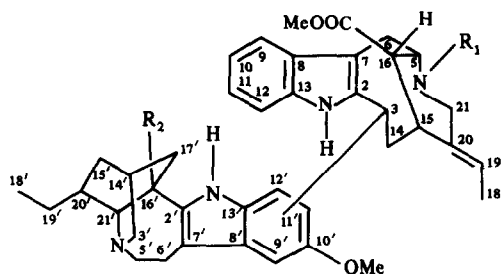
7 R₁ = Me, Z = O, R₂ = CH₂OH, R₃ = H

8 R₁ = Me, Z = αH, βOH, R₂ = H, R₃ = CO₂Me

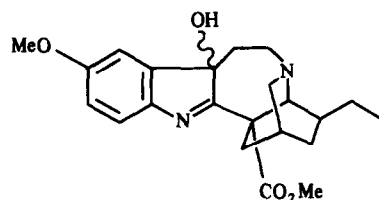


2 R₁ = CH₂OH, R₂ = H

3 R₁ = COOMe, R₂ = CH₂OH



1a	R ₁ = Me, R ₂ = COOMe,	LINKAGE at 11'
1b	R ₁ = Me, R ₂ = H,	" " 11'
1c	R ₁ = H, R ₂ = COOMe,	" " 11'
1d	R ₁ = Me, R ₂ = COOMe,	" " 9'
1e	R ₁ = Me, R ₂ = COOMe, 19-20 DIHYDRO,	" " 11'
1f	R ₁ = H, R ₂ = COOMe, 19-20 DIHYDRO,	" " 11'



5

at 60 and 100 MHz were obtained using TMS as int standard ¹³C NMR spectra were recorded at 252 MHz with Fourier transform using CDCl₃ as solvent and TMS as int standard MS were determined at 70 eV Silica gel 0.05-0.25 mesh (Carlo Erba) and silica gel HF₂₅₄₋₃₆₆ nm (Merck) were used for CC and TLC, respectively Detection of components was made by UV (254 and 305 nm) and spraying with Dragendorff's reagent followed by MeOH-H₂SO₄ and heating the plates at 150° for 5 min

Plant material Stem bark of *P. fuchsiae* (DC) Miers was collected in the Zeferino Vaz University City The air-dried bark (2.929 g) was moistened with a saturated NaHCO₃ soln and extracted in a Soxhlet with Et₂O On concn, the Et₂O extract gave a viscous oil which was added to a 10% HOAc soln and kept at 5° overnight After filtration the aq phase was extracted with Et₂O (extract A 6.65 g) and CHCl₃ (extract B 18.14 g) The pH was then raised to 8 with a saturated NaHCO₃ soln and extracted with Et₂O (extract C 8.52 g) and CHCl₃ (extract D 0.25 g)

Extract A (2.96 g) was fractionated on a silica gel column eluting with CHCl₃, and CHCl₃ with increasing amounts of MeOH, yielding voacamine (1a) (0.076 g) [9], voacangine (4) (0.41 g) [10] and voacanginehydroxyindolenine (5) (0.1388 g) [11] TLC also indicated the presence of demethylvoacamine (1c) [12] and affinisine (2) [13]

Extract B A Sephadex LH 20 column (6 g) eluted with CHCl₃-MeOH (9:1) permitted a crude separation of the dimeric from the monomeric compounds The combined fractions were further purified using a silica gel column and/or prep TLC, yielding the bisindoles voacamine (1a) (0.196 g) [9], demethyl-

voacamine (1c) (0.232 g) [12], decarbomethoxyvoacamine (1b) (0.178 g) [14] and voacamidine (1d) (0.206 g) [12], and the indoles vonchapotine (3) (0.221 g) [13] and affinisine (2) (0.329 g) [13]

Extract C (3 g) was fractionated on a silica gel column eluting with CHCl₃, and CHCl₃ with increasing amounts of MeOH, leading to the isolation of pervine (6) (0.1399 g) [8, 10], 16-epiaffinisine (7) (0.2043 g) [15], affinisine (2) (0.2167 g) [13] and decarbomethoxyvoacamine (1b) (0.1295 g) [14]

Dihydrovoacamine (1e) An EtOH soln of voacamine (1a) (0.206 g) with a catalytic amount of PtO₂ was submitted to hydrogenation (45 psi H₂) in a Parr apparatus for 30 min Filtration through a Celite pad and evapn of solvent furnished 0.198 g of dihydrovoacamine, mp 210-212° (MeOH), [α]_D²⁵ +40.9° (0.010 g/ml CHCl₃), UV λ_{max}^{CHCl₃} nm (log ε) 226.2 (4.75), 286.2 (4.28), 293.7 (4.28), IR ν_{max}^{CHCl₃} cm⁻¹ 3450, 1710; ¹H NMR δ 0.93 (m), 2.47 (3H, s), 2.60 (3H, s), 3.63 (3H, s), 3.93 (3H, s), 7.48 (1H, s), 7.60 (1H, s), MS m/z (rel int) 720 (55), 706 [M]⁺ (31), 511 (100)

Demethyldihydrovoacamine (1f) An EtOH soln of demethylvoacamine (1c) (0.206 g) was hydrogenated as above to furnish 0.149 g 1f Mp 210° (MeOH), UV λ_{max}^{CHCl₃} nm (log ε) 226.8 (4.68), 285 (4.23), 293 (4.22), IR ν_{max}^{CHCl₃} cm⁻¹ 3460, 1720; ¹H NMR δ 0.93 (t), 2.47 (3H, s), 3.67 (3H, s), 3.97 (3H, s), MS m/z (rel int) 720 (62), 706 [M]⁺ (44), 136 (100)

Acknowledgements—We wish to express our appreciation to Professor Anita J Marsaioli for helpful suggestions, Dr Norbert

Table 1 ¹³C NMR data for voacamine alkaloids and their derivatives

Carbon	4	8	9	7	1a	1b	1c	1e	1d	1f
2		135.4		134.4	135.5	135.5	135.5	135.6	134.9	135.7
3		66.8		189.5	37.4	37.6	37.1	37.2	37.1	37.3
5		59.4		56.5	59.9	59.7	53.3	59.2	59.8	53.0
6		19.6		19.3	19.8	19.4	24.6	19.2	19.0	25.5
7		107.3		120.4	109.6	110.0	110.2	109.8	110.8	110.9
8		128.7		128.1	129.4	129.4	129.7	129.6	129.9	129.5
9		117.6		121.2	117.2	117.0	117.1	117.2	116.9	117.3
10		118.6		120.2	119.8*	118.6	118.7	118.6	118.5	118.7
11		121.4		126.5	121.3	121.2	121.3	121.3	120.5	121.3
12		110.0		112.1	110.1	110.0	110.6	110.5	109.4	110.4
13		136.7		136.3	136.2	137.5	137.1	136.9	137.4	137.1
14		35.5		43.4	36.6	36.1	36.3	31.4	37.1	31.6
15		29.2		31.2	33.3	33.4	34.1	32.8	32.1	33.4
16		47.1		38.5	46.0	46.7	52.3	49.7	46.9	52.3
18		12.2		12.0	12.3	12.2	12.0	11.4	12.2	11.1
19		118.6		120.2	118.7*	118.6	117.3	23.5	118.5	23.5
20		136.5		135.2	137.7	137.8	140.1	43.7	137.9	45.9
21		53.9		51.9	52.3	52.2	44.1	51.9	52.3	40.8
COOMe		174.3		—	170.7	171.1	171.0	171.5	171.4	171.3
COOCH ₃		50.3		—	49.8	49.7	49.8	49.5	49.8	49.7
NMe		42.1		41.7	41.9	42.1	—	42.4	42.2	—
CH ₂ OH				66.7						
2'	137.3		142.9		136.9	142.1	137.6	137.5	138.8	137.5
3'	51.7		50.0		52.0	49.7	51.8	52.3	51.0	51.9
5'	53.1		54.2		53.0	54.1	53.0	53.0	53.9	53.0
6'	22.2		20.7		22.1	20.6	22.2	22.2	24.9	22.2
7'	110.0		109.1		109.6	108.4	109.6	110.3	109.0	109.8
8'	129.1		129.7		129.4	128.4	129.7	130.0	126.2*	130.1
9'	100.7		100.3		99.1	98.5	99.1	99.0	126.4*	99.2
10'	154.0		153.9		150.6	150.7	150.7	150.8	152.4	150.9
11'	111.9		110.8		127.1	127.9	127.1	127.0	112.5	127.1
12'	111.1		110.6		109.5	109.6	109.7	109.6	109.4	109.6
13'	130.6		130.0		130.0	129.0	130.1	130.3	131.8	130.2
14'	27.3		26.5		27.2	26.1	27.3	27.3	27.2	27.3
15'	32.0		32.0		31.8	31.7	31.9	31.9	31.8	31.9
16'	55.0		42.0		54.8	40.9	54.8	54.8	55.8	54.9
17'	36.5		34.2		36.2	33.9	36.3	36.4	33.5	36.4
18'	11.7		11.9		11.5	11.8	11.6	11.6	11.6	11.5
19'	26.7		27.8		26.6	27.6	26.7	26.7	26.7	26.7
20'	39.1		41.5		38.8	41.7	38.9	38.9	38.9	38.9
21'	57.6		57.5		56.9	57.5	57.0	57.0	58.2	57.0
COOMe	175.6		—		174.9	—	174.9	175.0	175.5	174.9
COOCH ₃	52.7		—		51.8	—	50.0	49.7	52.5	50.6
OMe	55.7		56.0		56.0	55.9	56.0	56.0	57.8	56.0

Spectra were obtained at 25.2 MHz in Fourier transform mode in CDCl₃ solutions. Chemical shifts are expressed on the TMS scale according to $\delta_{\text{TMS}} = \delta_{\text{CDCl}_3} + 76.9 \text{ ppm}$.

*Assignments for these signals within a vertical column may be reversed.

Neuss of Lilly Research Laboratories for providing an authentic sample of perivine and the Conselho Nacional de Desenvolvimento Científico e Tecnológico—CNPq—for financial support.

REFERENCES

- Kingston, D. G. I. (1978) *J. Pharm. Sci.* **67**, 272.
- Fernandez, M. E., Albonico, S. M. and Ruveda, E. A. (1967) *An. Asoc. Quim. Argent.* **55**, 239.
- Achenbach, H. (1966) *Tetrahedron Letters* 4405.
- Damak, M., Poupat, C. and Ahond, A. (1976) *Tetrahedron Letters* 3531.
- Wenkert, E., Cochran, D. W., Gottlieb, H. E., Hagaman, E. W., Braz F.º, R., Matos, F. J. A. and Madruga, M. I. L. M. (1976) *Helv. Chim. Acta* **59**, 2437.
- Ahond, A., Bui, A. M., Potier, P., Hagaman, E. W. and Wenkert, E. (1976) *J. Org. Chem.* **41**, 1878.
- Bombardelli, E., Bonati, A., Gabeta, B., Martinelli, E. M., Mustich, G. and Danieli, B. (1976) *J. Chem. Soc. Perkin Trans. 1*, 1432.
- Büchi, G., Manning, R. E. and Montu, S. A. (1964) *J. Am.*

- Chem Soc* **86**, 4631
- 9 Voticky, Z, Jahodar, L and Cava, M P (1977) *Collect Czech Chem. Commun* **42**, 1403
- 10 Holubek, J and Strouf, O (1965) *Spectral Data and Physical Constants of Alkaloids*, No 295, Heyden, London
- 11 Thomas, D W and Biemann, K (1968) *Tetrahedron* **24**, 4223
- 12 Achenbach, H and Schaller, E (1976) *Chem Ber* **109**, 3527
- 13 Achenbach, H (1966) *Tetrahedron Letters* 4405
- 14 Thomas, D W and Biemann, K (1965) *J Am Chem Soc* **87**, 5447
- 15 Naranjo, J, Pinar, M, Hesse, M and Schmid, H (1972) *Helv Chim Acta* **55**, 752